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# UTILITY PATENT APPLICATION TRANSMITTAL

(Only for new nonprovisional applications under 37 C.F.R. § 1.53(b))

Attorney Docket No. GRE-100C2

First Inventor or Application Identifier Alex E. S. Green

Title Process and Device for Pyrolysis of Feedstock

Express Mail Label No. EK318905314US

**APPLICATION ELEMENTS**

See MPEP chapter 600 concerning utility patent application contents.

1.  \* Fee Transmittal Form (e.g., PTO/SB/17)  
(Submit an original and a duplicate for fee processing)
2.  Specification [Total Pages 25]  
(preferred arrangement set forth below)
  - Descriptive title of the Invention
  - Cross References to Related Applications
  - Statement Regarding Fed sponsored R & D
  - Reference to Microfiche Appendix
  - Background of the Invention
  - Brief Summary of the Invention
  - Brief Description of the Drawings (if filed)
  - Detailed Description
  - Claim(s)
  - Abstract of the Disclosure
3.  Drawing(s) (35 U.S.C. 113) [Total Sheets 5]
4. Oath or Declaration [Total Pages 2]
  - a.  Newly executed (original or copy)
  - b.  Copy from a prior application (37 C.F.R. § 1.63(d))  
(for continuation/divisional with Box 16 completed)
    - i.  DELETION OF INVENTOR(S)  
Signed statement attached deleting  
inventor(s) named in the prior application,  
see 37 C.F.R. §§ 1.63(d)(2) and 1.33(b).

\*NOTE FOR ITEMS 1 & 13: IN ORDER TO BE ENTITLED TO PAY SMALL ENTITY FEES, A SMALL ENTITY STATEMENT IS REQUIRED (37 C.F.R. § 1.27), EXCEPT IF ONE FILED IN A PRIOR APPLICATION IS RELIED UPON (37 C.F.R. § 1.28).

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5.  Microfiche Computer Program (Appendix)
6. Nucleotide and/or Amino Acid Sequence Submission  
(if applicable, all necessary)
  - a.  Computer Readable Copy
  - b.  Paper Copy (identical to computer copy)
  - c.  Statement verifying identity of above copies

**ACCOMPANYING APPLICATION PARTS**

7.  Assignment Papers (cover sheet & document(s))
8.  37 C.F.R. § 3.73(b) Statement  Power of  
(when there is an assignee)  Attorney
9.  English Translation Document (if applicable)
10.  Information Disclosure Statement (IDS)/PTO-1449  Copies of IDS  
Citations
11.  Preliminary Amendment
12.  Return Receipt Postcard (MPEP 503)  
(Should be specifically itemized)
13.  Small Entity Statement(s)  Statement filed in prior application,  
(PTO/SB-09-12)  Status still proper and desired
14.  Certified Copy of Priority Document(s)  
(if foreign priority is claimed)
15.  Other: .....

16. If a CONTINUING APPLICATION, check appropriate box, and supply the requisite information below and in a preliminary amendment:  
 Continuation     Divisional     Continuation-in-part (CIP)    of prior application No. PCT / US98/16957

Prior application information: Examiner \_\_\_\_\_ Group / Art Unit: \_\_\_\_\_  
 For CONTINUATION or DIVISIONAL APPS only: The entire disclosure of the prior application, from which an oath or declaration is supplied under Box 4b, is considered a part of the disclosure of the accompanying continuation or divisional application and is hereby incorporated by reference. The incorporation can only be relied upon when a portion has been inadvertently omitted from the submitted application parts.

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02/18/00

Applicant or Patentee: Alex E.S. Green Attorney's  
Serial or Patent No.: \_\_\_\_\_ Docket No. GRE-100C2  
Filed or Issued: February 18, 2000  
For: Process and Device for Pyrolysis of Feedstock

VERIFIED STATEMENT (DECLARATION) CLAIMING SMALL ENTITY  
STATUS (37 CFR 1.9 (f) and 1.27 (b)) – INDIVIDUAL

As below named individual, I hereby declare that I qualify as defined in 37 CFR 1.9 (c) for purposes of paying reduced fees under section 41(a) and (b) of Title 35, United States Code, to the Patent and Trademark Office, with regard to the invention entitled Process and Device for Pyrolysis of Feedstock described in

- the specification filed herewith  
 application Serial No. \_\_\_\_\_, filed \_\_\_\_\_.  
 patent no. \_\_\_\_\_, issued \_\_\_\_\_.

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Alex E. S. Green  
NAME OF INDIVIDUAL Alex E. S. Green NAME OF INDIVIDUAL \_\_\_\_\_ NAME OF INDIVIDUAL \_\_\_\_\_  
Signature of Individual \_\_\_\_\_ Signature of Individual \_\_\_\_\_ Signature of Individual \_\_\_\_\_  
Feb. 18 2000 Date Date Date

DESCRIPTIONPROCESS AND DEVICE FOR PYROLYSIS OF FEEDSTOCK

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Cross-Reference to a Related Application

This is a continuation-in-part of co-pending application Serial No. PCT/US98/16957 filed August 14, 1998, which is a continuation-in-part of co-pending application Serial No. 08/912,485 filed August 18, 1997.

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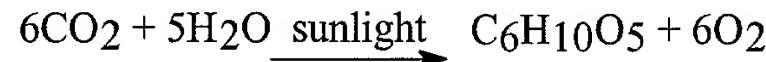
Background of the Invention

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Combustion of biomass has been used by humankind to generate heat and light for about 500 millennia. Biomass was the world's predominant energy source until fossil fuels took over in the industrial world, during the industrial revolution. As a renewable energy source, biomass can now re-assume greater energy loads while serving a number of other economic, environmental, and social purposes. Biomass can include, for example, wood, wood waste, agricultural waste, energy crops, municipal solid waste, sewage sludge and cellulosic type industrial waste.

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Biomass is made by photoreactions such as



leading to cellulosic type compounds such as  $\text{C}_6\text{H}_{10}\text{O}_5$ . Biomass fuel is greenhouse neutral since when burned it simply releases carbon dioxide and water that it took out of the atmosphere when it grew.

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The conversion of heat to mechanical work can be accomplished via hot air engines, steam engines or internal combustion engines (ICE). Hot air and steam can be made by the direct combustion of biomass. However, for use with reciprocating or turbine ICEs, which have been dominant in the 20<sup>th</sup> century, biomass is best converted to liquid or gaseous fuels.

Gasification is a thermal process of changing a solid fuel such as coal, biomass or

municipal solid waste into combustible gas and oil vapors. Four conventional biomass gasifier types that have evolved over many years are the fixed-bed updraft gasifier, the fixed bed downdraft gasifier, the moving bed gasifier, and the fluidized bed gasifier. Each type has advantages which are dependent upon the operating conditions, the output power required, and various other factors.

These conventional gasifiers typically create the heat for gasification by burning fuel in the gasification chamber itself. This can involve injecting air with its nitrogen into the gasification chamber, which dilutes the output gas with inert nitrogen as well as the products of combustion, including CO<sub>2</sub> and H<sub>2</sub>O. The fuel being burned to generate the heat for gasification is typically burned without sufficient oxygen, thus creating the fuel gas carbon monoxide.

A new class of gasifiers, referred to as indirectly heated gasifiers (IHGs), can generate at least a portion of the heat of gasification by combustion carried out in a separate chamber from the gasifier reactor. Accordingly, these IHGs reduce dilution of the output gas with nitrogen, carbon dioxide and water vapor. For example, one such gasifier heats sand in a separate combustion chamber and then transfers the hot sand into the gasification chamber to provide the heat for gasification. However, this process of heating and transferring sand is complicated and applies mainly to large gasification systems.

Depending upon the gasification agent and the gasifier, typical biomass gasifiers produce combustible CO, H<sub>2</sub>, CH<sub>4</sub>, and other light hydrocarbons, diluted with non-combustibles N<sub>2</sub>, CO<sub>2</sub> and H<sub>2</sub>O vapor. The heat of combustion of the product gas is determined by the biomass feedstock and gasification agent used, as well as by the operational conditions, such as pressure, temperature, residence time and heat loss or external heat input.

The types of gases produced by biomass gasification can be divided into three categories according to their heat value (HV). Low heating value (LHV) gas (~ 6 kJ/liter) is produced by traditional gasifiers when air is used as the gasifier agent. The gas is used on site since storage and/or transportation of LHV gas are not economically favorable. Medium heating value (MHV) gas (~ 13 kJ/liter) can be produced with traditional gasifiers when

oxygen is used as the gasifier agent since dilution by nitrogen is avoided. Medium HV gas can be used as fuel for internal combustion engines and gas turbines. In addition, medium HV gas can also be produced with IHGs in which the combustion chamber and the gasification chamber are separated. In IHGs the gasification process takes place without external oxygen (or nitrogen) and the output gas consists mostly of carbon monoxide with varying concentrations of the fuels hydrogen, methane, ethylene, ethane and other hydrocarbons, as well as some non-fuels such as carbon dioxide and water vapor. Medium HV gas can be used for the production of synthetic fuels, such as H<sub>2</sub>, gasoline, methanol, synthetic natural gas, etc. High HV gas (~ 37 kJ/liter) is usually produced from medium HV gases. These gases can be used as substitutes for natural gas which usually has a heating value of about 50 kJ/liter.

Gas turbines (GT) have shown promise as an efficient means of transforming heat into mechanical work and are now serving as major components of large new electricity generation systems using natural gas. For example, a low cost solid fuel (SF) cogenasifier fed by low cost local feedstocks can be coupled with smaller GT systems adapted for medium HV gas to produce electricity in, for example, remote regions where availability of electricity is limited. The cogenasification of biomass with other domestic fuels can provide a long term strategy for effective utilization of biomass. For example, the blending of oxygenated fuels such as biomass with carbonaceous fuels such as coals, coke, and chars in a small cogeneration system can have technological, economic, and environmental advantages. In addition, interest in distributed electricity generation is creating a need for low cost gasifiers.

Even with a low cost gasifier, a gasifier-microturbine generator (GMTG) might have difficulty these days competing economically with simpler natural gas microturbine generating systems in many locations because of the current low price of natural gas in the United States. However GMTGs could be economically competitive if they could also serve secondary and tertiary added value functions. The standby production of liquid fuels and chemicals from biomass is an example of a useful secondary function that could help in amortizing the capital cost of a new gasifier. Another societal function that could be served in the longer term by an IHG biomass gasifier is the gasification or liquefaction of the organic matter in metal laden biomass with the concentration of toxic metals is the char

ash. Examples of this need arise in the disposal of plants used in phytoremediation of toxic sites or the disposal of copper chromium arsenate (CCA) treated wood that has exceeded its lifespan. Additional functions, such as the use of the microturbine exhaust output for generating process steam or heat for drying crops or biomass fuel, would further enhance the 5 economic value of the GMTG system.

Brief Summary of the Invention

The subject invention pertains to unique and advantageous systems for gasifying and/or liquefying biomass. The systems of the subject invention utilize a unique design whereby heat from a combustion chamber or other source is used to directly gasify or liquify 10 biomass. In a preferred embodiment, the biomass is moved through a reactor tube in which gasification and/or liquefaction, i.e. the production of gases which condense at ordinary temperatures, takes place. In a specific embodiment, char is extruded from the biomass reactor tube and enters a combustion chamber where the char serves as fuel for combustion. The combustion chamber partially surrounds the reactor tube and is in direct thermal contact 15 with the reactor tube such that heat from the combustion chamber passes through the reactor wall and directly heats the biomass within the reactor tube. Other heat sources which can be utilized to provide the heat for gasification and/or liquefaction include, but are not limited to, electric oven, coal, heavy oil, tire crumb, electric tube furnace, microwave, and nuclear.

In a specific embodiment the subject invention provides methods and apparatus for 20 power generation utilizing a indirectly heated gasifier of the subject invention to provide fuel for small gas turbine generators. Thus, the system of the subject invention can be used for the gasification of solid fuel for small gas turbines.

The subject invention is also capable of serving secondary and tertiary added value 25 functions such as standby liquefaction of biomass or concentration of toxic metals in biomass into a residual char that represents a small percentage of the original biomass. The small percentage may be, for example, about 5% or less. The system of the subject invention can be operated in a variety of modes with a variety of feedstocks including non-biomass fuels such as coal, tire crumb, plastic chips, refuse derived fuel (RDF), and other organic matter.

An advantageous feature of the subject invention is its simple indirect heating of the 30 solid fuel which is to be gasified. The subject invention can utilize a simple, continuously

fed, high temperature reactor based upon advanced materials and coatings such as those developed for advanced gas turbines. Advantageously, the heat generated during the combustion stage of the power generation process can be conducted through the internal structure of the gasifier to heat the biomass which is introduced into the gasifier. There can exist a temperature gradient whereby the biomass is exposed to progressively higher temperatures as it moves through the gasifier and is heated to high temperatures for gasification and/or liquefaction.

In a specific embodiment of the subject invention biomass is conveyed through the system by at least one auger which rotates in such a way as to move the biomass from an inner hopper through the gasifier. Depending on output needs, multiple auger-reactors can be utilized. As the biomass gasifies, the gases can rise up through the biomass being moved by the auger blade such that the biomass can act as a filter for the output gases. In an alternative embodiment, the auger shaft can be hollow and comprise openings to the biomass so as to provide a means for gases and/or liquefaction particles produced from the gasification and/or liquefaction process to travel out through the hollow auger shaft, thereby bypassing the filtering action of the input feedstock.

A further aspect of the subject invention involves proper blending of fuels to insure proper hydrogen/carbon (H/C) and oxygen/carbon (O/C) ratios such that steam or other hydrogenating agents are not necessary.

A further aspect of the subject invention is a method and apparatus for operating to produce, for example, oil and/or liquids.

A further aspect of the subject invention is provisions for burning extruded char-ash-tar (CAT), or extruded char-ash (CA) when tar is fully gasified, with air to provide heat for the overall gasification or liquefaction processes.

A further aspect of the subject invention is the application of anaerobic indirectly heated gasifiers to the concentration of metals and other toxic substances in the char-ash. In this particular embodiment the char-ash burning step is by-passed and part of the derived gas and, when necessary, external fuel, electrical heater, or other heat source are used to provide the heat for the gasification or liquefaction processes.

A further aspect of the subject invention is a provision for adapting to a char-ash

capture and metal recovery mode, for example when the biomass contains contaminants.

A further aspect of the subject invention is a provision for adaptation to an ash encapsulation mode.

Other objects, advantages, and features of the subject invention will be apparent from the following description of the preferred embodiment taken in conjunction with the accompanying drawings.

#### Brief Description of the Drawings

**Figure 1** illustrates a cross-section of a solid fuel, indirectly heated, gasifier according to the subject invention.

**Figure 2** is a composite drawing illustrating important properties of various feedstocks.

**Figure 3** illustrates a cross-section of a gasifier in accordance with the subject invention, adapted to enhance production of oil and/or liquids.

**Figure 4** illustrates a cross-section of a gasifier in accordance with the subject invention, adapted to collect rather than burn, the char-ash.

**Figure 5** illustrates a cross-section of gasifier in accordance with the subject invention, incorporating a pressure vessel char-ash collector.

#### Detailed Disclosure of the Invention

The subject invention pertains to unique and advantageous systems for gasifying and/or liquefying biomass. The systems of the subject invention utilize a unique design whereby heat from a combustion chamber or other heat source is used to directly gasify or liquefy biomass. In a preferred embodiment, the biomass is moved through a reactor tube in which all the gasification and/or liquefaction takes place. In a specific embodiment, char exits the biomass reactor tube and enters the combustion chamber where the char serves as fuel for combustion. The combustion chamber or other heat source can partially surround the reactor tube and can be in direct thermal contact with the reactor tube such that heat from the combustion chamber or other heat source passes through the reactor wall and directly

heats the biomass within the reactor tube. Other heat sources which can be utilized to provide the heat for gasification and/or liquefaction include, but are not limited to, electric ovens, coal, heavy oil, tire crumb, electric tube furnace, microwave, and nuclear. The term "gasification" as used herein refers to, in general, the production of hot gases that do not condense at ordinary environmental temperatures and pressures. The term "liquefaction" as used herein refers to, in general, the production of hot gases that condense to liquids under ordinary environment temperatures and pressures. In accordance with the subject invention, gasification and liquefaction can occur simultaneously. The term "pyrolysis" as used herein refers to, in general, gasification and/or liquefaction.

A specific embodiment the subject invention provides methods and apparatuses for power generation utilizing an indirectly heated gasifier for small gas turbine generators. The subject invention can be used in solid fuel gasification for small gas turbines. The subject invention can also be utilized by providing multiple systems for medium size turbines.

Figure 1 illustrates a compact gasifier 1 which can produce, for example, a medium HV gas with an output capable of running a microturbine generator. The subject gasifier can utilize a variety of blended feedstocks, for example, wood chips, newspaper, mixed waste paper, peat, energy crops, agricultural residues as well as coal, tire crumb, plastics, RDF, lignite coal, lignin, bark, and other organic material. The output of a single or multiple module gasifier according to the subject invention can be sufficient to power a small gas turbine generator, for example as a part of a distributed electrical system. For example, a plurality of reactor tubes can be utilized to increase the throughput of feedstock. In a specific embodiment, six reactor tubes and six corresponding rotating augers can be utilized to increase the throughput of feedstock. The subject invention can take advantage of high temperature materials and coatings of the nature of those used for advanced gas turbines. Further, the subject gasifier can deliver medium HV gaseous fuels, for example free of ash, char, and tars, to a small gas turbine-generator. These small gas turbines can produce, for example, about 10 to 250 kW, as compared to gas turbine generators in the megawatt or multi-megawatt range.

An important feature of the subject invention is a means for feeding the material from

the inner hopper **10** into the reactor tube **2**. For example, a uniform or tapered auger **4** can be utilized. The auger **4** and reactor tube **2** should be able to withstand high temperatures. Metallic materials such as Haynes 214 or 230 alloys or high temperature Inco or Rolled alloys can be used such that the subject gasifier can accomplish high yield gasification or liquefaction in a single stage. Alternatively the reactor tube can be made out of advanced ceramics or composites. The higher temperatures not only foster rapid and complete volatilization, but also promote the chemical reactions with the char, largely carbon (C):



Advantageously, these chemical reactions increase the heating value of the gaseous output, further increase volatile generation, and decrease char residue.

In a specific embodiment, the subject gasifier involves the feeding of biomass into inner hopper **10** such that a rotating auger flight **5** can force the biomass into reactor tube **2**. While residing in the inner hopper, the biomass can be heated by exposure to the bottom cone of the inner hopper and the gases emanating from the reactor tube **2**. Once in the reactor tube **2**, the biomass can be further and more rapidly heated by heat produced by, for example, a flame in the combustion chamber **6** which is conducted through the reactor tube walls **3**. This heat conducted through the reactor tube walls **3** causes gasification of the biomass as the biomass is moved through the reactor tube **2** by the rotating auger flight **5**. The gases given off during gasification can travel through the biomass in reactor tube **2** toward the inner hopper **10**, so as to be filtered by the incoming feedstock, and subsequently outputted via, for example, an output gas opening **13**. Once the biomass reaches the end of the reactor tube **2**, it becomes a char-ash-tar (CAT) with the amount of the tar component depending upon feed rate, temperature of operation, input feedstock, and other variables.

The subject invention can provide a combustion chamber **6** for burning the extruded char-tar with air to provide at least a portion, and preferably all, of the heat and corresponding high temperatures needed for the gasification or liquefaction process within the reactor tube **2**. The heat from the combustion chamber **6** conducts through the wall **3** of the reactor tube **2** and can be transferred to the feedstock by contact of the feedstock with the reactor tube **2**. A temperature gradient can be established, for example in steady state, which

provides higher temperatures as the biomass heats up as it moves along the reactor tube and is subsequently transformed into the char-ash-tar plug **16**. At the end of the reactor tube, the char-ash-tar can be pushed out of outlet **50** and enter the combustion chamber **6**. Once in the combustion chamber **6**, air, for example preheated, can be provided by blower **9**, and the char or char-ash-tar can be burned to create heat that is then conducted through the reactor tube walls to gasify the biomass. The air can be preheated by, for example, the combustion chamber's exhaust gases using a heat exchanger.

Advantageously, the subject invention can utilize this indirect heating of feedstock for oxygen-free volatilization of the feedstock. The outer surface **40** and **41** of the gasifier **1** can be made of combinations of materials which insulate the internal chambers of the gasifier from the outside. This keeps the heat, which is used to heat the biomass, in the gasifier and reduces the heat in the surrounding environment. The reactor tube **2**, preferably tapered, can be made from materials and coatings which allow heat from combustion chamber **6** to conduct through to the biomass. In order to allow for gasification of the biomass, the materials used for the reactor tube walls, the auger, and the auger blade should, preferably, be able to withstand high temperatures. Higher gasification temperatures can increase gas yield. In a preferred embodiment, the gasification temperature exceeds about 800° C (1472° F) and most preferably is about 1000° C (1822° F) to about 1200° C (2191° F). Preferably, production of condensable gases, i.e. liquefaction, can occur within a temperature range of about 400° C (752° F) and about 800° C (1472° F), and most preferably within the range of about 500° C (932° F) and about 600° C (1112° F).

The gasification and/or liquefaction of the biomass can create pressure in the reactor tube **2** and inner hopper **10**. This pressure can range greatly depending on the exit flow rate of the gasifier. For example, the pressure in the inner hopper can be maintained as low as about 1 psi gauge. In another specific embodiment, pressures on the order of about 100 psi can be generated in the reactor tube **2** and maintained in the inner hopper **10**. A means for the pyrolysis-gasification gas output to exit upward through the incoming feedstock allows the feedstock to serve as a filter, capturing tar comprising condensable complex organics which can then make an additional pass through the high temperature reactor to be broken

down into smaller molecules. It is important to provide a means for preventing the gasification gases from exiting the reactor tube into the combustion chamber **6**, wherein the combustion chamber **6** is preferably at a lower pressure. The CAT plug **16** at the outlet of the auger-reactor can provide a means for keeping gases from the reactor tube **2** from leaking into the combustion chamber **6**. Accordingly, the CAT plug **16** insures that the gases generated in the auger-reactor work up through the incoming feedstock to the inner hopper. Advantageously, condensable gases, tars, liquids, and particles captured by the incoming feedstock make additional passes through the reactor tube, allowing these bigger molecules to be broken down into lighter gases.

It is important that the heat conducted from the combustion chamber through the reactor tube walls **2** be able to reach and, therefore heat, the biomass in the reactor tube. Depending on the type of biomass and other parameters, the reactor tube can be, for example, on the order of one to several inches in diameter in order to allow sufficient heating of the biomass. In a preferred embodiment, the diameter of the reactor tube is between about 1 and about 5 inches. Materials and coatings currently used in advanced gas turbines can be used in the subject gasifier, particularly in critical components such as the reactor tube.

In a preferred embodiment, the subject gasification method involves blending fuels in such a way as to insure proper H/C and O/C ratios, thus rendering steam or another hydrogenating agent unnecessary. Blending of feedstocks in accordance with the subject invention also gives acceptable char-ash-tar residue properties to form a CAT plug **16** capable of withstanding pressures in the reactor tube **2** and inner hopper **10** of approximately 1-100 psi. In addition, the blending is adjusted to achieve a net heating value input and sufficient char-tar output to provide the heat required for endothermic gasification processes. This blending can occur in, for example, the outer hopper **8**. In a specific embodiment, the system of the subject invention comprises a means for taking chipped feedstock from an outer hopper **8** at atmospheric pressure to an inner hopper **10** at gas pressures up to about 100 psi. For example, a lock hopper **12** can be utilized to input feedstock to the gasifier. Alternatively, other high pressure solid feeders such as rotary valves as in the recently developed Stamet POSIMETRIC system can be adapted for this application.

Referring to Figure 3, in a preferred embodiment, the outlet **50** at the bottom of the reactor tube **15** is shaped such that the CAT forms a plug **16** capable of withstanding differential pressures up to about 100 psi or more. This plug **16** insures that the product gas goes toward the inner hopper **10** and the combustion gases go out, for example, the combustor exhaust **14**. In a specific embodiment, the combustion exhaust can proceed through opening **43** in a flame guard plenum **42**, where the exhaust can heat the funnel wall **44** carrying feedstock to the auger-reactor. This can begin the process of raising the feedstock to high temperatures and, in addition, extract heat from the combustion exhaust gases. Further extraction of the heat of the combustion exhaust gases can be accomplished by heating the incoming air **51**, **52** using a heat exchanger.

At outlet **50** of orifice **17**, an additional plugging means, for example an adjustable or spring loaded cone-like plug **18**, can be used to plug orifice **17**, for example at start up, and can be opened by the extrusion pressure when an adequate CAT plug **16** is formed. The plug **18** can also prevent output gases from leaking into the combustion chamber **6** and prevent combustion gases from leaking up into the reactor tube. This plug **18** can, for example, be operated by or maintained by spring loading **23**. In these ways the outlet **50** can be adjustable, to optimize the CAT extrusion to serve as a plug for any feedstock.

As the CAT plug **16** is extruded out of outlet **50**, the char falls onto, for example, grate **7**. Blower **9** can blow heated air, into the combustion chamber **6** to provide over-fire tangential air **51** for combustion of the char, and under grate **7** to provide under-fire air **52** for combustion of the char. In a specific embodiment, the exhaust gases from **14** can be used with a simple heat exchanger to preheat the air entering the combustion chamber. Once the char is burned, the ashes can fall through grate **7** and spider **19** into ash receptacle **11**. A start-up gas burner **20** (e.g., propane) can be utilized for initial heating of the reactor tube and can be located, for example, integrally with spider **19**.

In a specific embodiment, a burning means, for example comprised of pipes **53** and valves **32**, can be used for burning some of the product gas through burner **55** in the combustion chamber **6** to provide heat for gasification. In addition, an external source of gaseous fuel can be provided through input **31** for the combustion chamber when, for

example, the feedstock has insufficient heating value, the feedstock is wet, or when the char ash is to be collected because of toxic content.

In an alternate embodiment, as shown in Figure 5, grate 7 is replaced by a solid plate so that chamber 11 becomes a simple pressure vessel char-ash collector. In this embodiment, upon entering the high temperature zone the feedstock, in the absence of oxygen, begins to pyrolyze, forming gases, liquids (tars), and a solid residue (char/ash). As the feedstock reacts, the solid extrudes out the bottom into the char collector 11. The gases and vapors that are generated flow upwards, through the reactor and the incoming feedstock in the hopper, where some of the condensable gases (tars) are removed and some pass through, depending upon the feedstock height in the hopper. The output of the gasifier is passed into separator 27 that can perform various functions depending upon the mode of operation of the microgasifier. In the gas production mode the separator primarily separates particles from the gas stream that have not already been filtered out by the incoming feedstock. In this mode molecular sieves can be used to separate molecular hydrogen from the remaining gases which can exit through orifice 29 into a compressor for storage and premium fuel use. These molecular sieves may be, for example, zeolite or ceramic.

In the liquid production mode, i.e., the production of condensable gases, separator 27 provides a water-cooled condensing system or a refrigerator cooled system to condense the oils and liquids, including water, arising from lower temperature pyrolysis of the incoming feedstock. These liquids can exit through orifice 33 and be collected during low gas needs of the microgasifier. Occasionally, these liquids can be distilled on site or be sent, for example, to a local distiller for separation into premium liquid fuels or chemicals. In the pressure vessel receptacle mode gases, such as steam, carbon dioxide, or natural gas, can be injected through input 31 to enhance carbon conversion in the reactor via the reactions  $H_2O + C \rightarrow H_2 + CO$ ,  $CO_2 + C \rightarrow 2CO$  or  $C + CH_4 \rightarrow C_2H_4$ .

The subject methodology can be used to capture metals chemically or by being physically bound in the solid feedstocks. This can be useful with respect to waste disposal problems and can provide a basis for developing and using IHG/L technology even when, in the absence of this service, it is not economically competitive in power generation. Such

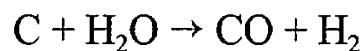
needs can include a). the capture of lead, mercury, arsenic or other toxic metals found in construction and demolition debris (C&DD), b). the disposal of old CCA treated wood, c). the disposal of plants grown on mined phosphate lands, d). the disposal of plants used for phytoremediation, e). the concentration of the metals picked up by plants used for phytomining, f). the "de-alkalization" (removal of potassium and/or sodium) of plant matter to produce gas for co-firing biomass derived fuel in coal boilers so as to avoid slagging and fouling problems.

For materials laden with highly refractory metals, pyrolysis at temperatures below the boiling point should leave most of the metals in the char-ash. Steam, carbon dioxide, air oxygen, or natural gas injected at the bottom of the reactor tube or through holes in the auger pipe-shaft can at high temperatures convert char and tar into useful gases or liquids. For materials such as CCATW, with its semi-volatile metal arsenic, the post gasification system can bear at least part of the clean-up burden. For example, the gas output from pyrolysis can be scrubbed by one or more of a variety of techniques known in the art to remove volatile metals such as arsenic. Cleaning the gas or liquid before using it for fuel has a major advantage over a post combustion clean-up system. A post combustion system needs to process some 5 or so times the weight or volume of gas that such a pre-combustion system would process. Furthermore, if applied to an IHG output, the sensible heat that must be taken out of the precombustion gas is small compared to the heat content of the gas. This is not the case if one scrubs the gas from an air blown gasifier that generates a much lower heating value gas. Accordingly, with medium heat value gases heat losses can be tolerated better than when scrubbing low heating value gases from air blown gasifiers and from combustion stack gases.

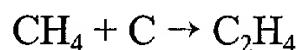
For feedstocks laden entirely with volatile metals the gas clean up system should bear most of the clean-up burden. Here, very high temperature processing can insure fast and thorough volatile metal purging of the solid fuel. Good heat recovery is preferred so that not too much sensible heat is wasted. A IHG in accordance with the subject invention can provide useful advantages in that the output gas can give up most of its heat to the incoming feedstock, serving the double function of cooling and filtering the gas to an extent

controllable by adjusting the level of feed in the inner hopper.

One of the earliest motivations for working on cutilization of domestic fuels was the possibility that there would be advantageous synergies that could be exploited to help in oil displacement. Biomass can be used to assist the processing of coal in the subject IHG/L to achieve conditions that can remove and capture arsenic, mercury or other volatile metals from the coal. Biomass blends with lignite coal that is closest from an H/C, O/C standpoint to lignin or bark can be used. The purged lignite can be used to fuel the process, although some carbon can remain in the ash. Steam can be used to hydrogenate the carbon to take advantage of the reaction:



Temperatures which can overcome the activation barrier can be used, leading to the possible hydrogenation reaction



This reaction can use natural gas (mainly  $CH_4$ ), another fossil fuel, to convert solid carbonaceous coal into a high energy gaseous form suitable for turbines. The ethylene can be used as feedstock for polyethylene production.

Dual service carbon sequestering schemes can also be utilized. For example,  $CO_2$  can be used as an oxidizer with lignite coal as feedstock. For example, consider the reaction



Here  $C_6H_5O$  is a surrogate for lignite coal and  $C_6H_5O_2$  or  $C_7H_5O_3$  is a surrogate for lenoardite or humic acid that has a high value as a soil amendment. This reaction can go rapidly at high temperatures. If desired, a catalyst can be used to promote it. In addition to consuming  $CO_2$ , the reaction can produce valuable products, such as leonardite and/or humic acid. Advantageously, coal itself can be used as the scrubber, addressing the  $CO_2$  problem while producing useful products with good market value.

A means for adjusting feed rate to meet the desired gas power output can also be provided. The feed rate for a desired output can be estimated, for example when firing with dry biomass. Figure 2 is a composite diagram giving the H/C and O/C atomic ratios of various potential solid feedstocks. Also shown in Figure 2 are the heating values relative to

carbon (32 kJ/g) and the volatile percentage assigned by ASTM weight loss method. An empirical rule that 1 gram of biomass at about 1000°C typically produces about 0.75 liters gas with a heating value of approximately 15 kilojoules/liter can be used. Accordingly, one gram gives about 11 kilojoules gas energy. Hence, one gram/sec is capable of producing 11 kilowatts gas heating power. Assuming a 30% capability of converting gas fuel to electrical energy, which may be considered typical, gives 3.3 kW. Accordingly, a feed rate of one gram/sec corresponds to 3600 gram/hr or  $3.6 \text{ kg/hr} = 3.6 \times 2.2 \text{ (lb/kg)} = 7.9 \text{ lb/hr} = 3.3 \text{ kWe}$ . Therefore a rule of thumb is  $= 7.9/3.3 = 2.4 \text{ lb/hr}$  biomass gives 1 kWe, for typical dry biomass. This corresponds to a heat rate of about 17M BTU/kW which is comparable with the heat rates of large but older coal or oil utilities. This heat rate can be further improved by optimizing the fuel blend utilized and/or the operational variables of the system.

Note that since the biomass heat value (HV) per gram is about  $0.6 \times 32 \text{ kJ} = 19 \text{ kJ}$  whereas the HV of the gas produced is about 15 kJ, about 4 kJ are expended in the gasification process. If 20% of the initial feedstock produces a char-tar that has a HV of about 30 kJ/g the heat value of about 6 kJ is available for each gram of feedstock. This is more than enough char-tar to provide the energy required for gasification.

The corresponding calculation for other feedstocks can be carried out in an analogous way. For any feedstock, the solid fuel heating value per gram must be greater than the heating value of the gas generated per gram and must produce a char tar residue which when combusted will provide the endothermic energy needed for gasification. As referred to above, blending of available fuels may be conducted as described herein to achieve optimal performance for a given application. When blending fuels, attention must be paid to the physical and textural properties of the constituents to insure the filtering action of the input feedstock and the plugging effectiveness of the char-tar-ash extrusion.

In a specific embodiment of the subject invention, if single auger throughput is not sufficient, a multiple auger reactor system driven by, for example, a single variable speed motor and comprising a single inner hopper and a single combustion chamber can be implemented.

The basic microgasifier of the subject invention can be adapted to produce, for

example, oil and/or liquids. Referring to Figure 3, an inner tube **21** can be provided, for example, with apertures **54** that can be opened or closed, for example by rotating inner tube **21** up to or through 90°, to provide paths for the pyrolysis and gasification gases to go directly out of the auger reactor region to the top of the inner hopper **10**. These gases can exit the inner tube through aperture **26** and subsequently through output gas opening **13**. An external demister and water cooled condenser can be provided in the separator **27** to cool the condensable gases in order to extract oils and/or liquids. The non-condensable gases can be burned in the combustion chamber through outlet **28** and valve **32** with burner **55**. Alternatively, through high temperature plumbing **56**, part of these non-condensable gases can be pumped through the inner tube to speed the transport of the condensable gases to the condenser separator **27**. Additionally, a means for adjusting excess air and tangential swirl can be used in the combustion chamber in order to lower the temperature in the combustion chamber and, therefore, lower the temperature acting on the reactor tube. In a specific embodiment, further reduction of the temperature acting on the reactor tube can be accomplished by deflecting the flame zone from the tube. The feed rate can also be increased so that the average temperature experienced by the feedstock before escaping as gases through apertures **54** is lowered.

The basic microgasifier of the subject invention may also be modified or adapted in order to gasify feedstock with toxic contents. An anaerobic (without air) gasifier can lower the formation of volatile metal oxides or otherwise lower the volatilization of metallic constituents in favor of deposition in the char-ash residue. Referring to Figure 4, an embodiment in accordance with the subject invention which can be used to gasify feedstock with toxic contents is shown. The gasification reactor is anaerobic, which can lower the formation of volatile metallic oxides or otherwise promote metallic deposition in the char-ash. For example, in the embodiment shown in Figure 4, the change from Figure 3 is the replacement of the male cone plug **18** by a female hollow tube **49** that extrudes char-ash, for example, through the diagonal orifices at its bottom. The openings of the ring burner-spider **19** are covered to avoid burning the char-ash laden with toxics. Advantageously, this facilitates the adaption of the gasifier for the disposal of discarded copper chromium arsenate

(CCA) treated wood and subsequent reconcentration of the CCA, and/or its constituents. This embodiment can also be useful for the disposal of plant matter used in phytoremediation to capture toxic metals from contaminated sites. The char-ash-tar outlet is modified to feed into a collector for later processing.

5       The gasifier consists of three major components that can be assembled, for example, with the help of the positioning rings **36** or flanges. The upper assembly is the pressure capable inner hopper and combustion exhaust plenum-outlet. The middle assembly is the combustion. The bottom assembly is the char-ash collector from the ring burner downward.

10      A means for inputting external solid feed (e.g., coal and/or tire chips) to provide heat of gasification or to enrich the feedstock can also be added to the outer hopper. In addition, an external hopper **15** and system for feeding stored fuel (e.g., coal, tire chips, RDF, heavy oil) directly to the combustion grate can be provided. The feeding of stored fuel directly to the combustion grate can provide gasification energy when, for example, the char-ash is laden with toxics and, therefore, not burned. In addition, the feeding of solid fuels directly to the combustion grate can be useful when the normal feedstock is wet and/or when external gas supplies are limited. Feeding biomass to the gasifier via the outer hopper **8** and coal to the combustion via this external hopper can also have operational advantages when biomass feedstock must be stretched.

15      In a specific embodiment, a co-utilization system can be implemented in which a high density fuel or energy source such as coal, heavy oil, tire crumb, an electric oven, microwave source, electric tube furnace, nuclear source can provide energy for thermally converting biomass to more valuable gaseous liquid or even solid (e.g., metallurgical charcoal, activated charcoal, etc.) form. This embodiment can have a variety of applications and can be used in a number of ways. In this dual feedstock mode, or even the standard mode, chemicals can be blended with the biomass, for example, in the outer hopper, to neutralize unwanted constituents, or to catalytically promote desirable constituents, in the output gases or liquids. A means for such blending can also be provided. Advantageously, such blending can enhance the utility of the output gas or liquid for energy generation, medical applications, for food process applications or to make solvents, colorants, disinfectants, fumigants or other

specialty chemicals.

In addition, a specific embodiment can comprise a means for char-ash glassification. In a specific embodiment, low melting temperature material is added to the feedstock so that the extruded char-ash is melted and forms a glassy char-ash. In an additional embodiment, alkali rich materials such as grasses are included in the feedstock to yield a “wet” char-ash.

It should be understood that the examples and embodiments described herein are for illustrative purposes only and that various modifications or changes in light thereof will be suggested to persons skilled in the art and are to be included within the spirit and purview of this application and the scope of the appended claims.

Claims

1        1. A process for pyrolysis of feedstock, comprising the following steps:  
2              introducing feedstock into, and moving said feedstock through, a reactor tube; and  
3              heating the feedstock within said reactor tube to a sufficient temperature such that pyrolysis  
4              occurs,

5              wherein the feedstock is introduced into the reactor tube from an inner hopper,  
6              wherein gases of pyrolysis travel through the feedstock in the inner hopper such that said  
7              feed stock in the inner hopper acts as a filler,

8              wherein heat for heating the feedstock is generated by a heat source selected from the  
9              group consisting of: combustion chamber, gases, electric oven, coal, heavy oil, tire crumb,  
10             electric tube furnace, microwave, solar, and nuclear.

1        2. The process according to claim 1, wherein said feedstock comprises a substance  
2              selected from the group consisting of biomass wood chips, newspaper, mixed waste paper,  
3              peat, energy crops, agricultural residues, coal, tire chips, plastics, RDF, and other organic  
4              matter.

1        3. The process according to claim 1, wherein the heat generated by the heat source  
2              is conducted to the feedstock within the reactor tube through a reactor tube wall.

3        4. The process according to claim 1, wherein the feedstock is moved through the  
4              reactor tube by a rotating auger.

1        5. The process, according to claim 1, wherein pyrolysis occurs within a temperature  
2              range from about 800°C (1650°F) to about 1200°C (2190°F) such that substantially  
3              anaerobic gasification occurs.

1        6. The process according to claim 1, wherein pyrolysis occurs within a temperature

2 range from about 400°C (752°F) to about 800°C (1472°F) such that liquefaction occurs.

1       7. The process according to claim 1, further comprising the steps of: introducing  
2 feedstock into, and moving said feedstock through, at least one additional reactor tube;  
3              and heating the feedstock within said at least one additional reactor tube to a  
4 sufficient temperature such that pyrolysis occurs;  
5              wherein the feedstock is introduced into at least one additional reactor tube from the  
6 inner hopper.

1       8. The process according to claim 1, wherein said reactor tube comprises an exit  
2 orifice, feedstock exiting the reactor tube via the exit orifice enters a pressure vessel such that  
3 the pressure from the pressure vessel controls the flow of gases of pyrolysis from exiting into  
4 the pressure vessel.

1       9. The process according to claim 1, wherein a gas is injected into the reactor tube.

1       10. The process according to claim 9, wherein said gas is selected from the group  
2 consisting of CO<sub>2</sub>, steam, natural gas, oxygen, and air.

1       11. The process according to claim 4, wherein the auger comprises a hollow shaft  
2 having at least one opening, wherein gases of pyrolysis can exit through said hollow shaft.

1       12. The process according to claim 10, further comprising the step of controlling the  
2 flow of the gas into the reactor tube in order to adjust the conversion of char and tar exiting  
3 the reactor tube into useful gases and/or liquids.

1       13. The process according to claim 1, further comprising the step of capturing the  
2 feedstock residue exiting the reactor tube, wherein said process is useful for pyrolysis of

3 feedstock comprising a contaminant.

1       14. The process according to claim 1, further comprising the step of capturing the  
2 feedstock residue exiting the reactor tube, wherein said process is useful for pyrolysis of  
3 feedstock used for phytomining.

1       15. The process according to claim 13, wherein said contaminant is selected from  
2 the group consisting of heavy metals, lead, mercury, highly refractory metals, volatile metals,  
3 copper, chromium, arsenic, copper chromium arsenate and other toxics.

1       16. A device for pyrolysis of feedstock, comprising:  
2           a reactor tube within which pyrolysis of feedstock occurs;  
3           a means for moving feedstock through the reactor tube;  
4           a means for heating the feedstock within said reactor tube to a sufficient temperature  
5 such that pyrolysis occurs; and  
6           an inner hopper, wherein the feedstock enters the reactor tube from the inner hopper,  
7 wherein gases of pyrolysis travel through the feedstock in the inner hopper such that said  
8 feedstock in the inner hopper acts as a filter.

1       17. The device according to claim 16, wherein the means for moving said feedstock  
2 through said reactor is a rotating auger.

1       18. The device according to claim 16, wherein said reactor tube comprises an exit  
2 orifice, wherein the feedstock residue exiting the exit orifice enters a pressure vessel, wherein  
3 the pressure from the pressure vessel controls the flow of gases exiting the exit orifice.

1           19. The device according to claim 16, wherein the auger comprises a hollow shaft  
2 having at least one opening, wherein the gases of pyrolysis can exit through said hollow  
3 shaft.

1           20. The device according to claim 18, further comprising a means for injecting a gas  
2 into the exit orifice of the reactor tube.

1           21. The device according to claim 20, wherein said gas is selected from the group  
2 consisting of carbon dioxide, steam, natural gas, oxygen, and air.

1           22. The device according to claim 16, further comprising:  
2           a means for capturing the feedstock residue exiting the reactor tube, wherein said  
3 device is useful for pyrolysis of feedstock containing contaminants.

1           23. The device according to claim 16, further comprising:  
2           a means for capturing the feedstock residue exiting the reactor tube, wherein said  
3 device is useful for phytomining.

1           24. The device according to claim 19, comprising a means whereby a portion of the  
2 pyrolysis gases or external gases are injected into a lower end of the hollow shaft to hasten  
3 the transport of condensable gases to an external condenser liquid separator.

1           25. A method of pyrolysis of feedstock containing at least one contaminant using  
2 indirectly heated gasification, comprising the following steps:  
3           moving feedstock containing at least one contaminant through a reactor tube; and  
4           heating the feedstock within said reactor tube to a sufficient temperature such that  
5 pyrolysis occurs,  
6           wherein low oxygen conditions of pyrolysis leads to lower containment of the at least  
7 one contaminant in the gaseous output and higher capture and concentration of the at least

8 one contaminant in the feedstock residue.

1        26. The method according to claim 25, wherein said at least one contaminant is  
2 selected from the group consisting of heavy metals, lead, mercury, refractory metals, volatile  
3 metals, copper chromium arsenate, copper, chromium, arsenic and other toxics.

1        27. The method according to claim 25, further comprising the step of scrubbing gas  
2 output from the pyrolysis to remove volatile metals from the gas output.

1        28. The method according to claim 25, wherein low oxygen conditions of pyrolysis  
2 leads to essentially anaerobic pyrolysis, wherein said essentially anaerobic pyrolysis lowers  
3 the formation of volatile metallic oxides and promotes metallic deposition in the feedstock  
4 residue.

1        29. The method according to claim 25, wherein said method is used for disposal of  
2 plant matter used in phytoremediation.

1        30. The method according to claim 25, wherein the biomass is introduced into the  
2 reactor tube from an inner hopper, and wherein gases of pyrolysis travel through biomass in  
3 the inner hopper such that said biomass in the inner hopper acts as a filter.

1        31. A method for recovery of a substance residing in plant matter used in  
2 phytoremediation, comprising the following steps:

3              moving plant matter used in phytoremediation through a reactor tube; heating the  
4 plant matter within said reactor tube to a sufficient temperature such that pyrolysis occurs;  
5 and capturing plant matter residue exiting the reactor tube,

6              wherein a substance residing in the plant matter is recovered in the captured plant  
7 matter residue.

1           32. A method for recovering of a substance residing in plant matter used in  
2 phytomining, comprising the following steps:

3                 moving the plant matter used in phytomining through a reactor tube; heating the plant  
4 matter within said reactor tube to a sufficient temperature such that pyrolysis occurs; and  
5 capturing plant matter residue exiting the reactor tube,

6                 wherein a substance residing in the plant matter is recovered in the captured plant  
7 matter residue.

1           33. A process for pyrolysis of feedstock, comprising the following steps:

2                 introducing feedstock into, and moving said feedstock through, a reactor tube; and  
3 heating the feedstock within said reactor tube to a sufficient temperature such that pyrolysis  
4 occurs.

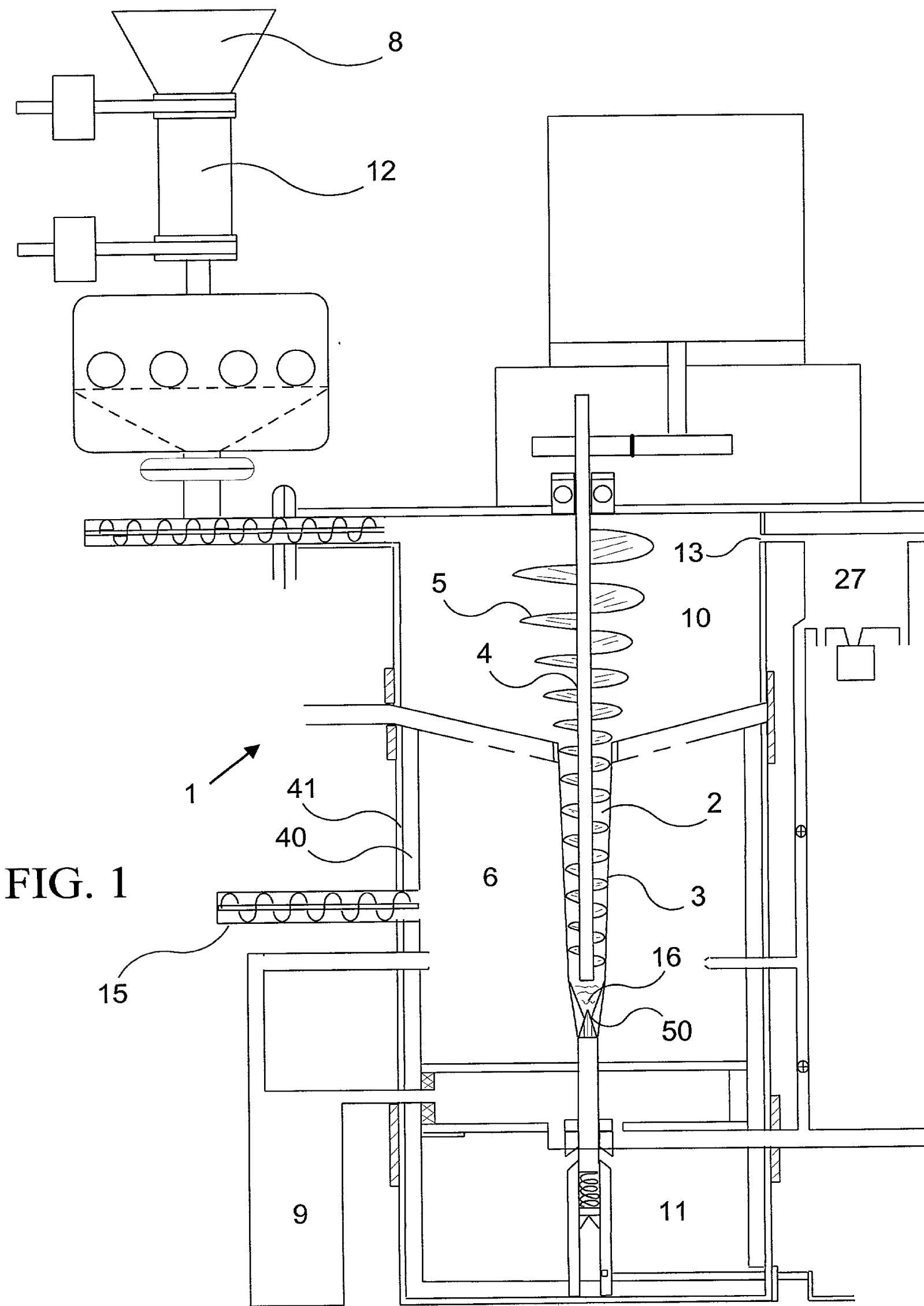
1           34. The process, according to claim 33, further comprising the step of introducing  
2 a gas into the reactor tube.

1           35. The process, according to claim 34, wherein the gas is selected from the group  
2 consisting of: CO<sub>2</sub>, steam, natural gas, oxygen, and air.

1           36. The process, according to claim 33, wherein the reactor tube comprises an exit  
2 orifice, feedstock exiting the reactor tube via the exit orifice enters a pressure vessel such that  
3 the pressure from the pressure vessel controls the flow of gases of pyrolysis exiting into the  
4 pressure vessel.

Abstract

The subject invention pertains to unique and advantageous systems for gasifying and/or liquefying biomass. The systems of the subject invention utilize a unique design whereby heat from a combustion chamber is used to directly gasify or liquefy biomass. In  
5 a preferred embodiment, the biomass is moved through a reactor tube in which all the gasification and/or liquefaction takes place. Preferably, char exits the biomass reactor tube and enters the combustion chamber where the char serves as fuel for combustion. The combustion chamber partially surrounds the reactor tube and is in direct thermal contact with the reactor tube such that heat from the combustion chamber passes through the reactor wall and directly heats the biomass within the reactor tube.  
10



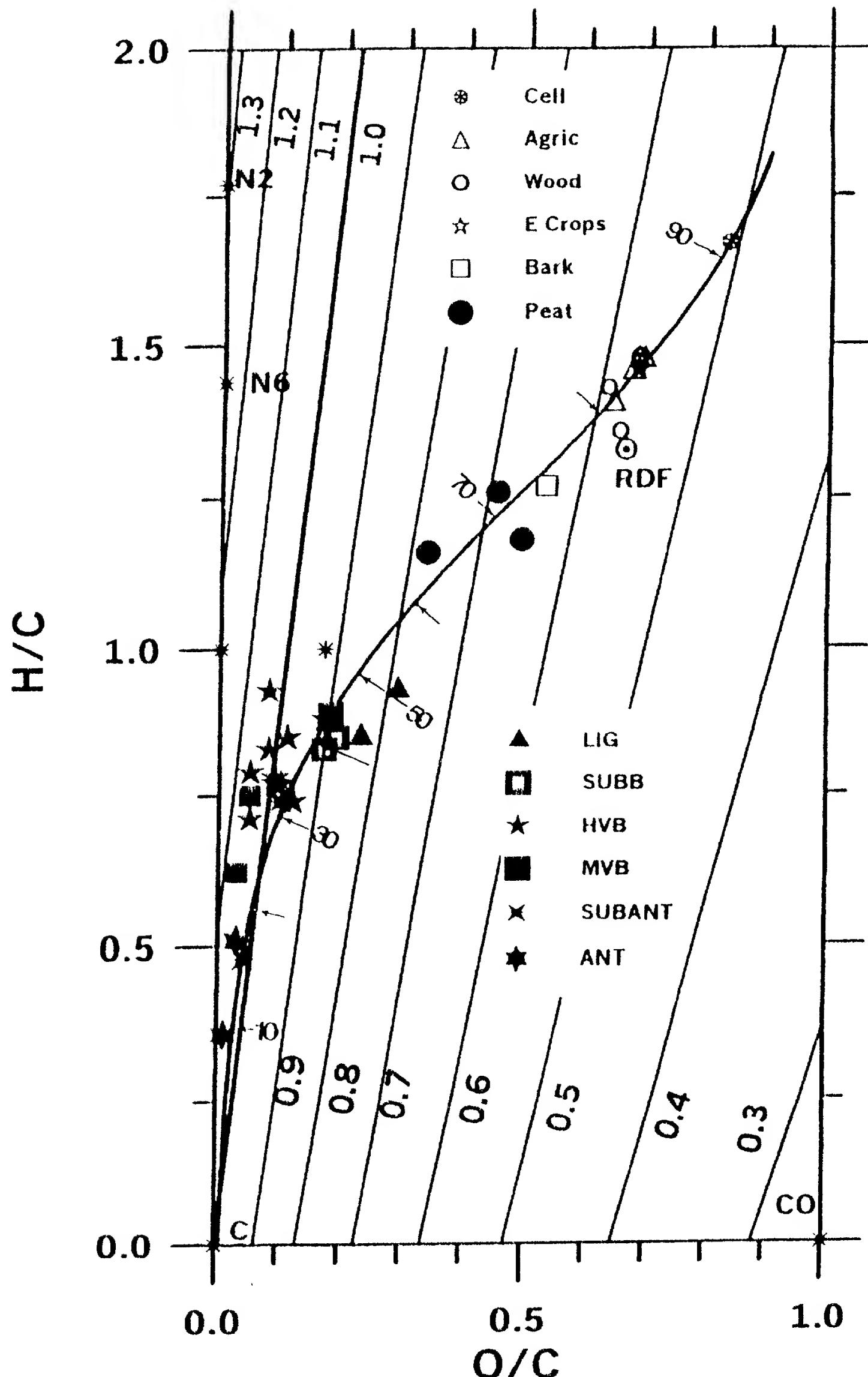


FIG. 2

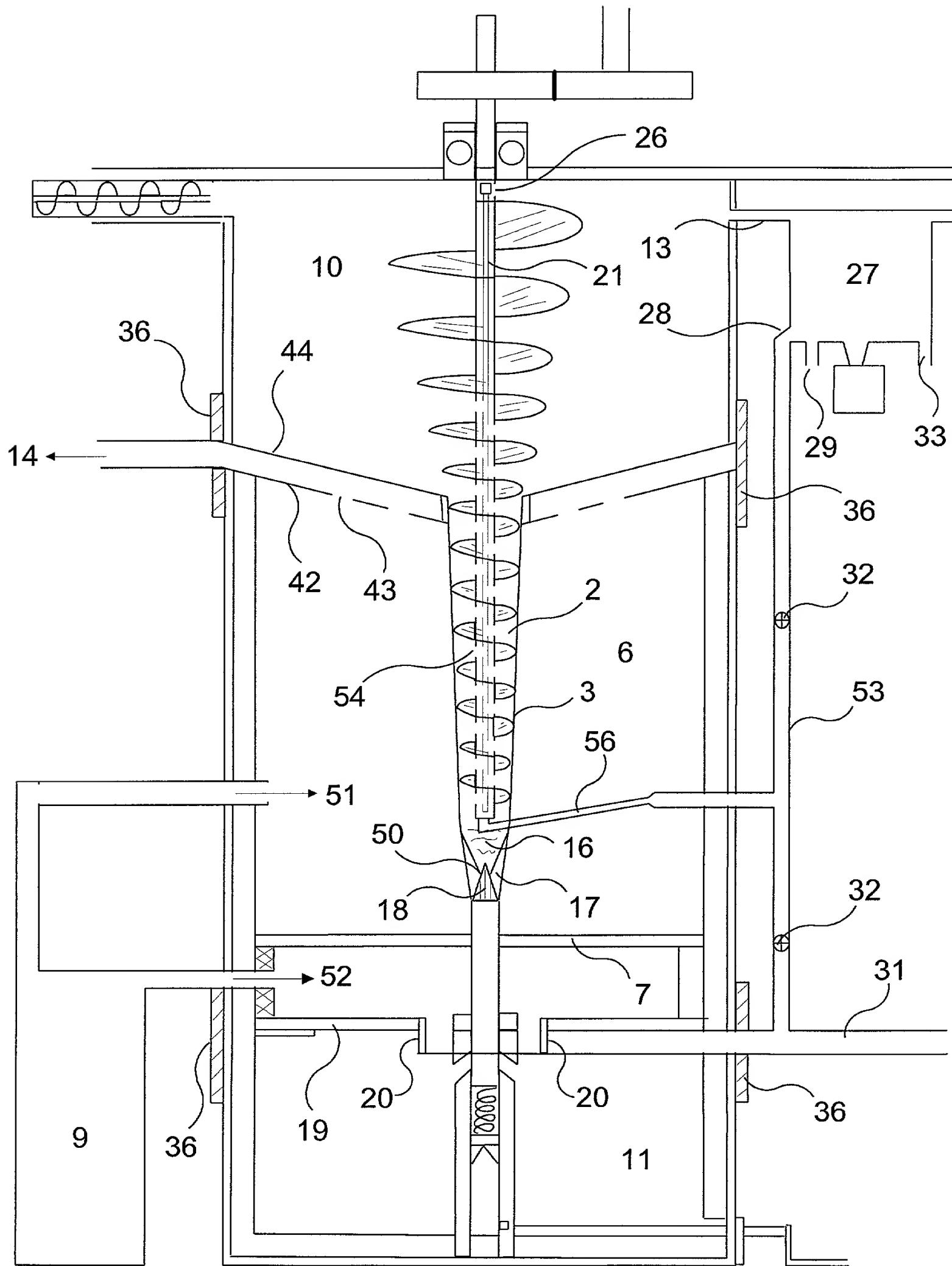


FIG. 3

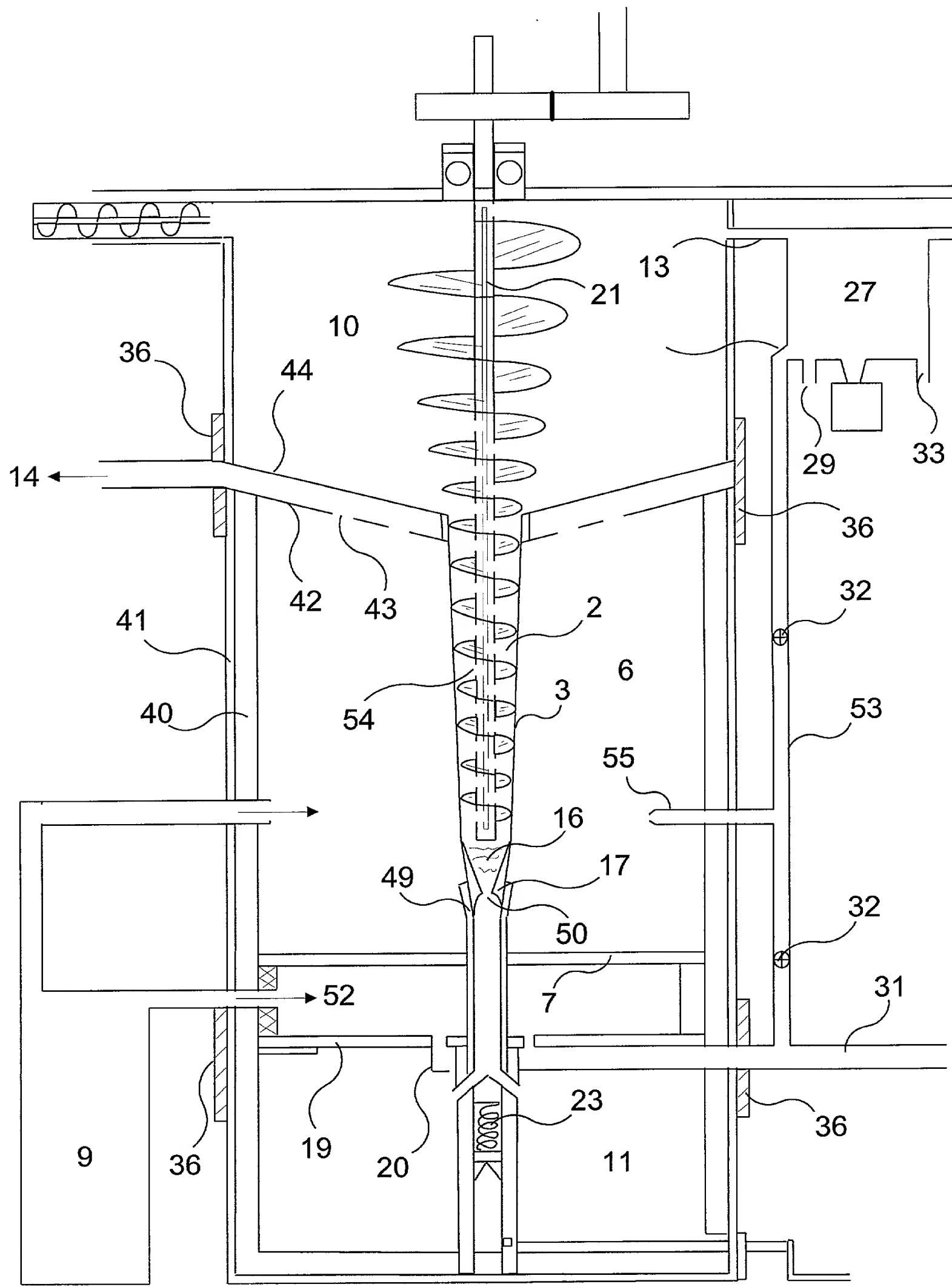


FIG. 4

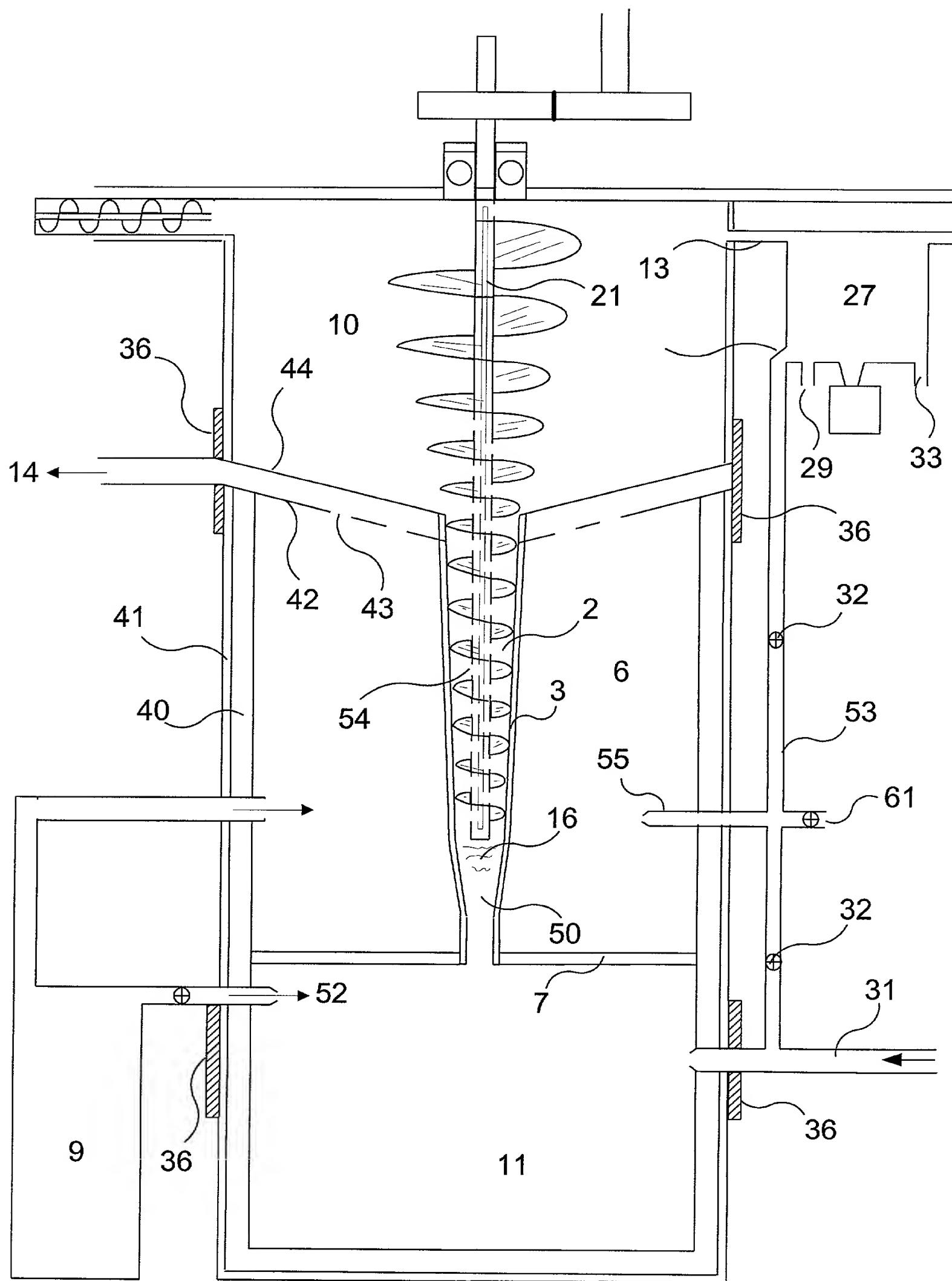


FIG. 5

## DECLARATION (37 CFR 1.63) AND POWER OF ATTORNEY

As a below-named inventor, I hereby declare that:

My residence, post office address, and citizenship are as stated below next to my name; and

I believe that I am the original, first, and sole inventor (if only one name is listed below), or an original, first, and joint inventor (if plural names are listed below) of the subject matter which is claimed and for which a patent is sought on the invention entitled **Process and Device for Pyrolysis of Feedstock**, specification for which

is attached hereto.

was filed \_\_\_\_\_, Serial No. \_\_\_\_\_.

I hereby state that I have reviewed and understand the contents of the above-identified specification, including the claims, as amended by any amendment referred to above.

I acknowledge the duty to disclose information which is material to the examination of this application in accordance with Title 37, Code of Federal Regulations, §1.56(a).

I hereby claim foreign priority benefits under Title 35, United States Code §119 and/or §365 of any foreign application(s) for patent or inventor's certificate listed below and have also identified any foreign application for patent or inventor's certificate having a filing date before that of the application on which priority is claimed:

Application Serial No.	Country	Filing Date	Priority Claimed
PCT/US98/16957	PCT	14 August 1998	Yes

I hereby claim priority benefits under Title 35, United States Code §119 of any provisional application(s) for patent listed below:

Application Serial No.	Filing Date	Priority Claimed
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I hereby claim the benefit under Title 35, United States Code, §120 and/or §365 of any United States application(s) listed below and, insofar as the subject matter of each of the claims of this application is not disclosed in the prior United States application(s) in the manner provided by the first paragraph of Title 35, United States Code, §112, I acknowledge the duty to disclose material information as defined in Title 37, Code of Federal Regulations, §1.56(a) which became available between the filing date of the prior application and the national or PCT international filing date of this application:

Application Serial No.	Filing Date	Status (Patented, Pending, Abandoned)
08/912,485	August 18, 1997	Pending

I hereby declare that all statements made herein of my own knowledge are true and that all statements made on information and belief are believed to be true; and further that these statements were made with the knowledge that willful false statements and the like so made are punishable by fine or imprisonment, or both, under Section 1001 of Title 18 of the United States Code, and that such willful false statements may jeopardize the validity of the application or any patent issued thereon.

I hereby appoint the following persons registered to practice before the Patent and Trademark Office as my attorneys with full power of substitution and revocation to prosecute this application and all divisions and continuations thereof and to transact all business in the Patent and Trademark Office connected therewith: David R. Saliwanchik, Reg. No. 31,794; Jeff Lloyd, Reg. No. 35,589; Doran R. Pace, Reg. No. 38,261; Christine Q. McLeod, Reg. No. 36,213; Jay M. Sanders, Reg. No. 39,355; James S. Parker, Reg. No. 40,119; Frank C. Eisenschenk, Reg. No. 45,332; Jean Kyle, Reg. No. 36,987; Seth M. Blum, Reg. No. P-45,489.

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Residence \_\_\_\_\_ Citizenship \_\_\_\_\_

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\_\_\_\_\_ Date \_\_\_\_\_

Signature of Third Joint Inventor

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Name of Fourth Joint Inventor \_\_\_\_\_

Residence \_\_\_\_\_ Citizenship \_\_\_\_\_

Post Office Address \_\_\_\_\_

\_\_\_\_\_ Date \_\_\_\_\_

Signature of Fourth Joint Inventor